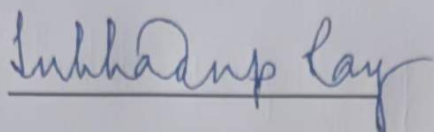


NOTICE

Date: 6.6.2022

This is hereby notified that 4th semester B.Com Program students are to submit the following assignments, in respect of their Compulsory English – ‘Appreciating Poetry’ (ENGLC 401) course, positively by 28.06.2022 at the English Department office. The assignment should be hand written on plain A4 size paper:

1. Features of the Sonnet
2. “Daffodil” as a nature poem.



Coordinator, Dept of English,
Bidhan Chandra College, Asansol

Name : Sheikh Nadim
Roll No : 1032004113051058
Registration No. : KNU20103003458
Phone No : 6294982657
Date : 28.06.2022

B.Com Semester IV Program (2022) Internal.

Subject Discipline : Compulsory English
Subject Name : Appreciating Poetry (English)
Subject Code : ENGLC401

Qn 1 :- What is Sonnet?

Answer :- Traditionally, the Sonnet is a fourteen-line poem written in iambic pentameter, employing one of several rhyme schemes, and adhering to a tightly structured thematic organization. The name is taken from the Italian sonetto, which means "a little sound or song".
Discover more poetic terms • Types of Sonnets.

Salient Features of Sonnet :-

- 1) They are fourteen lines long.
- 2) The fourteen lines are divided into four subgroups.
- 3) The first 3 subgroups have four lines each, which makes them "quatrains" with the 2nd and 4th lines of each group containing rhyming words.
- 4) The sonnet then concludes with a two-line subgroup, and these 2 lines rhyme with each other.
- 5) There are typically ten syllables per line, which are phrased in a iambic pentameter.
- 6) All sonnets have 14 lines, which can be broken down into four sections called quatrains.
- 7) The rhyme scheme of a Shakespearean sonnet, for example, is ABAB / CDCD / EFEF / GG.
- 8) Sonnets are written in iambic pentameter, a poetic meter with 10 beats per line made up of alternating unstressed and stressed syllables.

Q 2:- Consider Daffodil as a Nature Poem.

Answer:- William Wordsworth, the most prominent poet of the Romantic age, is known as a nature-lover and a nature poet. The poem "Daffodils" is one of the finest examples of the expression of the poet's love of nature from the very first line of the poem to the finishing end, he shows this love in the first stanza of "Daffodils" Wordsworth compares himself to a cloud that floats high over valleys and hills. Then he goes on to describe how he came across a host of golden daffodils which are "fluttering and dancing in the breeze". He personifies the daffodils as if they were happy and dancing just like a human being would do, compares the flowers to the shining stars on the Milkyway, and compares their happiness with that of the waves in the lake beside them. Finally, the poet reflects on what joy the memory of the flowers bring to his mind when he is in a vacant or pensive mood.

The poet declares "A poet could not be gay / in such a jound company". He also uses "I gazed and gazed" to express how charmed he had been by the daffodils. And the final lines "And then my heart with pleasure fills / and dances with the daffodils" says it all. Moreover; flowers, lake, trees, valleys, hills, cloud, stars - mention of all these natural object is not a mere coincidence. This is deliberate attempt from the poet to write all these things that bring him joy, that makes him happy, that essentially suggests only one thing that The poet is a lover of Nature.

✓



Bidhan Chandra College

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Asansol – 713304, Dist – Paschim Bardhaman , West Bengal , Ph : 0341-2283020
Visit us: www.bccollegeasansol.ac.in; E. Mail: bccollege.office@gmail.com

Ref. No.: _____

Date: _____

DEPARTMENT OF CHEMISTRY

NOTICE (Session 2021-2022)

It is hereby informed to all VI-th Semester students that Assignments will be provided to all of you on 20-04-2022. All the students are asked to be present on that day.

Note: This is mandatory for all Vith Semester students.

Date: 12/04/2022

S. B. Khan
.....
for
Head
Department of Chemistry



Assignment 2022

Subject: Chemistry (Honours)

Semester : VI

1. Write a comprehensive note on Pinacol-Pinacolone rearrangement. Your note should contain basic introduction, mechanism, factors influence the rearrangement and its application. Your writing should be within 500 words.



Pinacol - Pinacolone Rearrangement

● **Presented By :-** Rina Chand

● **College :-** Bidhan Chandra College, Asansol

● **Course :-** B.sc (Hons) in Chemistry

● **Univ. Regn. No. :-** KU/REG/18/03000935 of
2018-19

Molecular Rearrangements

Molecular Rearrangements

- Rearrangement reactions involves the migration of a group or an atom from one center (migration origin) to another (migration terminus) - within the same molecule.
- Rearrangements are divided into intramolecular and intermolecular processes. In intramolecular process, the group that migrates is not completely detached from the system in which rearrangement is taking place. On the other hand, in intermolecular process, the migrating group is first detached and later reattached at another site.

● Rearrangements can be roughly classified on the basis of the migration origin and migration terminus.

- i) Rearrangements to Electron Deficient Carbon
- ii) Rearrangements to Electron Deficient Nitrogen.
- iii) Rearrangements to Electron Deficient Oxygen
- iv) Sigmatropic Rearrangements.
- v) Other Rearrangements.

● Rearrangements to Electron Deficient Carbon ●

→ Pinacol - Pinacolone Rearrangement

→ Wagner-Meerwein Rearrangement

→ Semjarov Rearrangement

→ Benzil - Benzoic acid Rearrangement

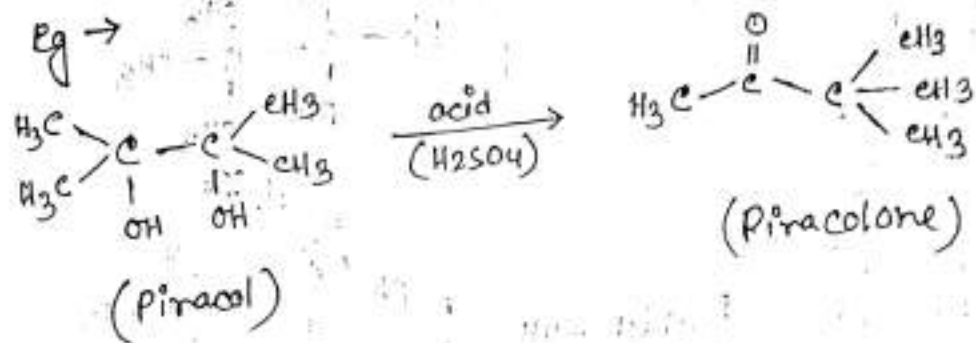
→ Wolff Rearrangement

→ Stenone - phenol Rearrangement

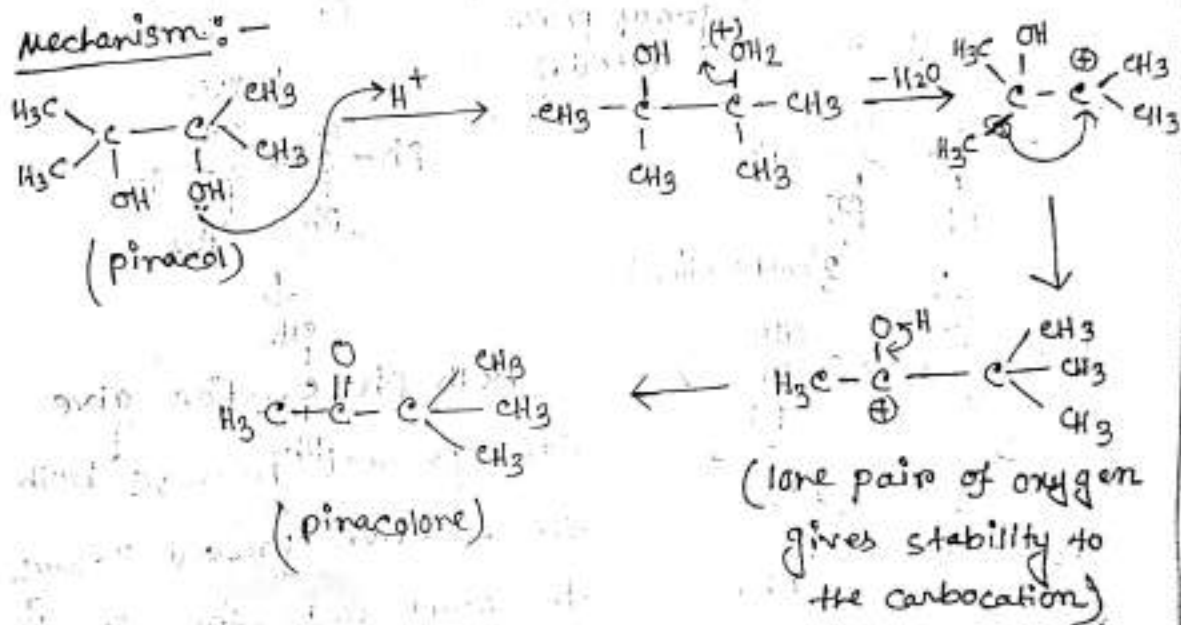
• Pinacol - Pinacolone Rearrangement •

vicinal diol-type compound, is called pinacol when it is treated with acid (H_2SO_4) a special type of rearrangement take place through carbocation intermediate giving carbonyl compound known as pinacolone and the rearrangement is termed as pinacol - pinacolone rearrangement.

The name is originated from the conversion of pinacol to pinacolone on treatment with acid.



Mechanism: -

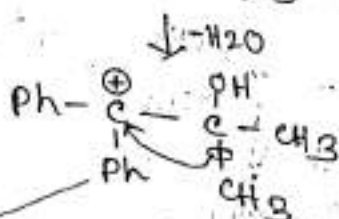
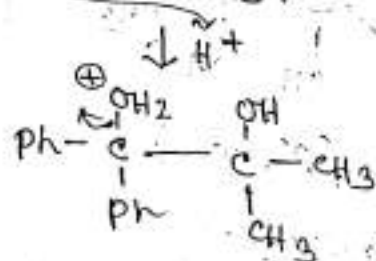
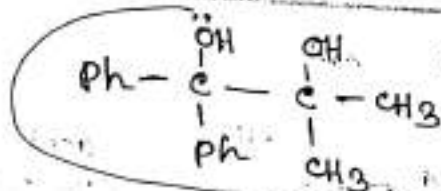
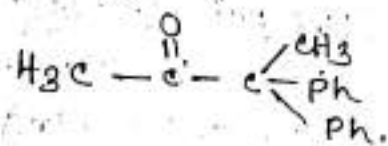
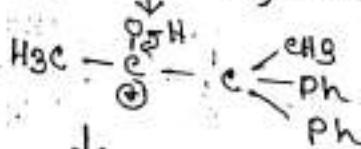
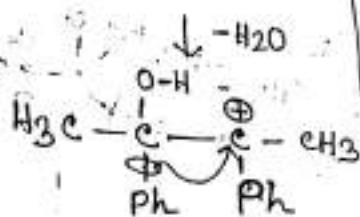
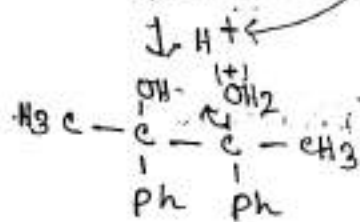
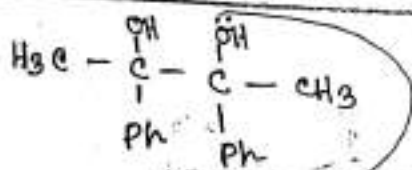


• Migratory aptitude:-

Aryl group > 3° alkyl group > 2° alkyl group > 1° alkyl group.

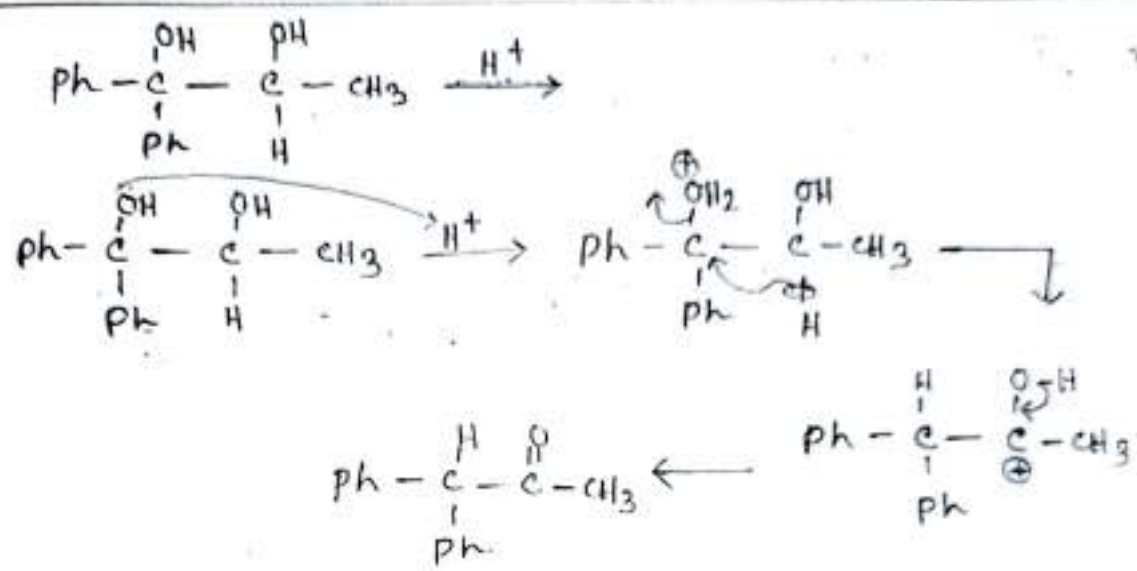
• Electron donating group at para and meta position in aryl system will increase the migratory aptitude.

• Electron withdrawing group at any position in aryl system will decrease the migratory aptitude.



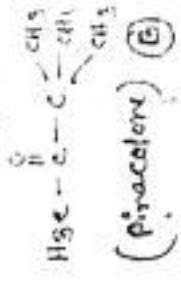
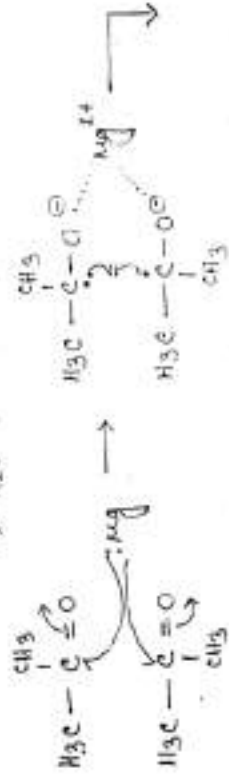
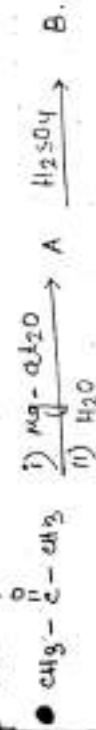
which -OH group protonated?

Both the reaction give same product because both the reaction proceed through the same carbocation intermediate.



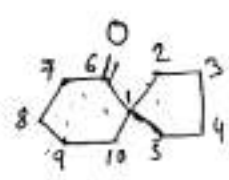
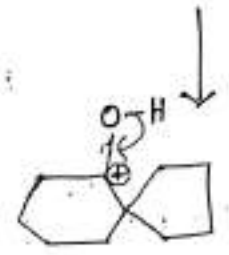
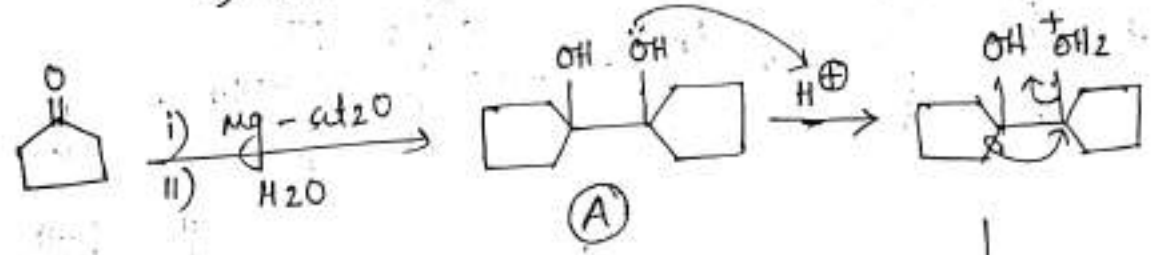
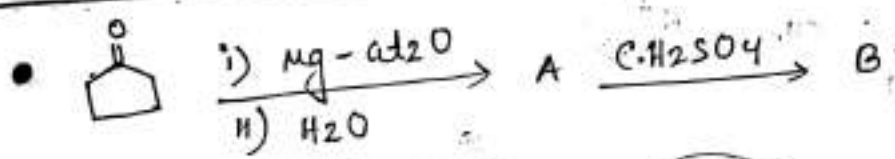
Hydrogen will migrate preferably than the other group due to steric factor.

How pinacol is prepared?



(pinacol)

(pinacolone)

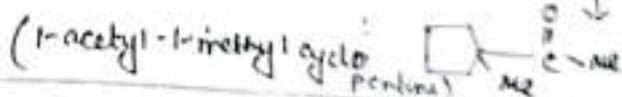
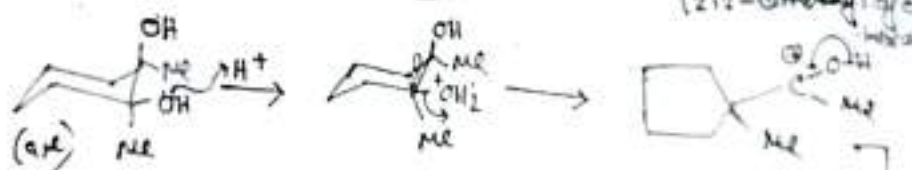
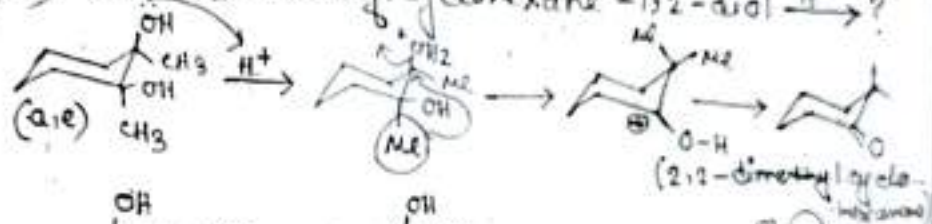


Spiro [4.5] Decan - 6 - one

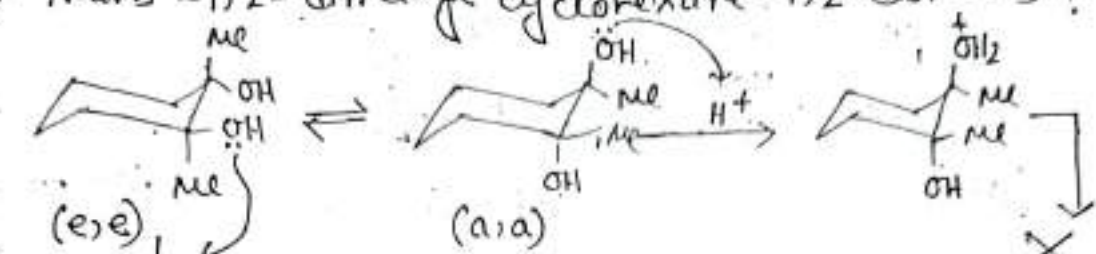
• Pinacol - pinacolone rearrangement in alicyclic system like cyclohexane :->

In cyclohexane system the group on bond which is situated anti to that of leaving group i.e H₂O will migrate preferably.

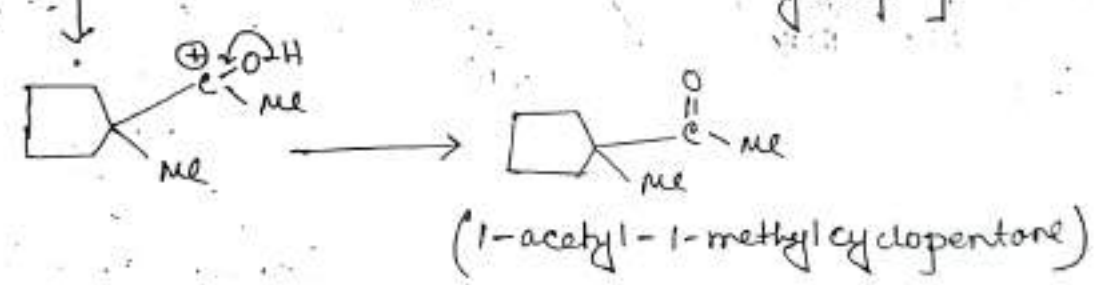
eg -> i) cis-1,2-dimethylcyclohexane-1,2-diol $\xrightarrow{H^+}$?



11) Trans-1,2-dimethylcyclohexane-1,2-diol $\xrightarrow{H^+}$?



[No reaction since -OH group never act as a migrating group]



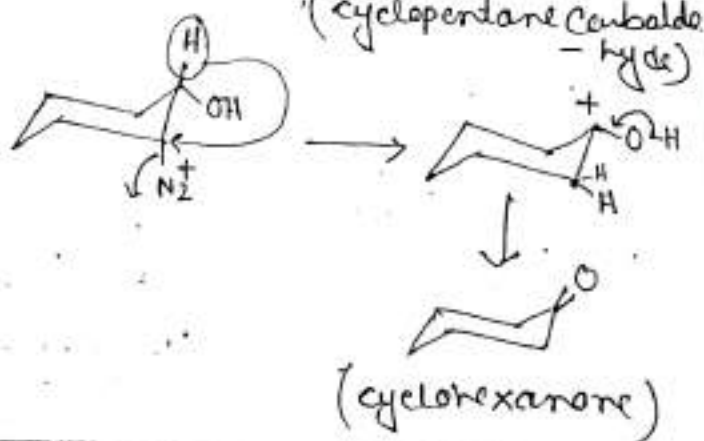
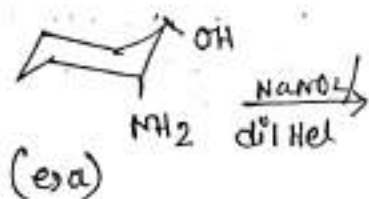
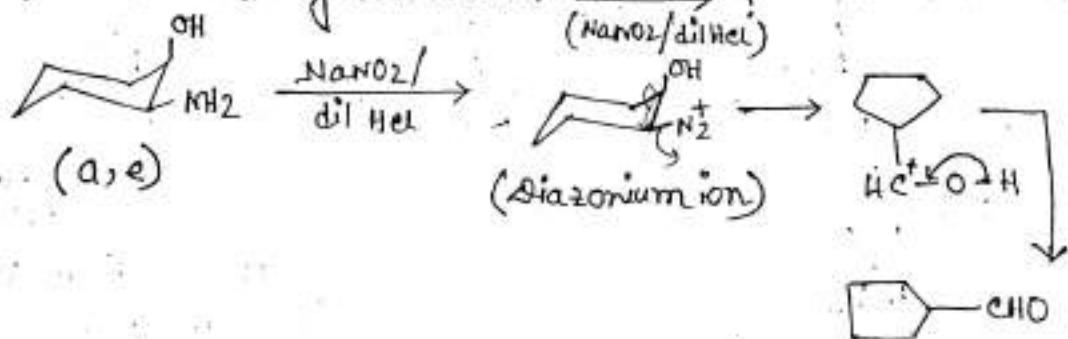
● Semi-pinacol-pinacolone rearrangement →

Some rearrangement are known which are mechanistically alike with pinacol-pinacolone rearrangement but the leaving group is different these rearrangement are termed as Semi-pinacol-pinacolone rearrangement.

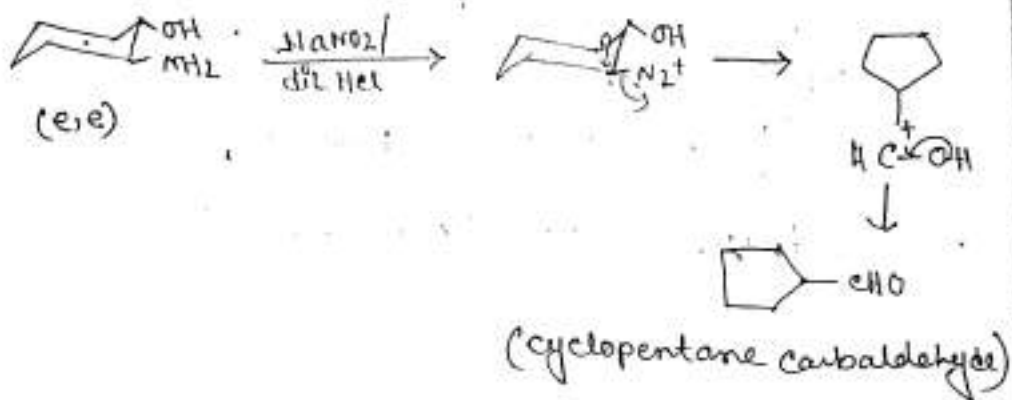
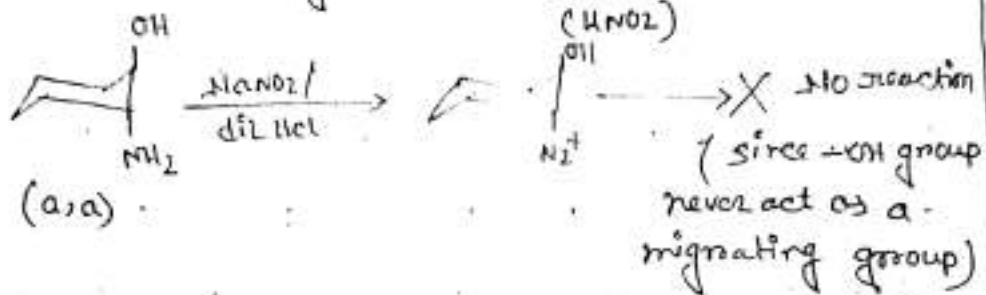
Eg → De-amination reaction

Reagent → HNO_2 ($+\text{HNO}_2 + \text{dil HCl}$)

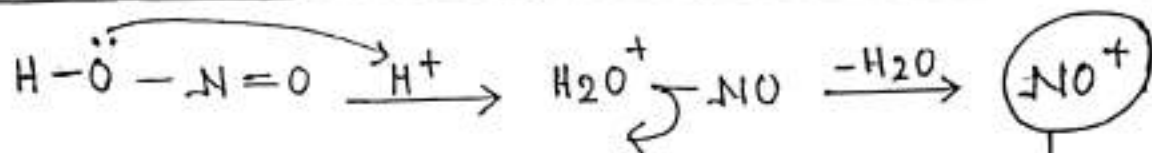
• cis-2-amino cyclohexanol $\xrightarrow[\text{(NaNO}_2/\text{dil HCl)}]{\text{HNO}_2}$?



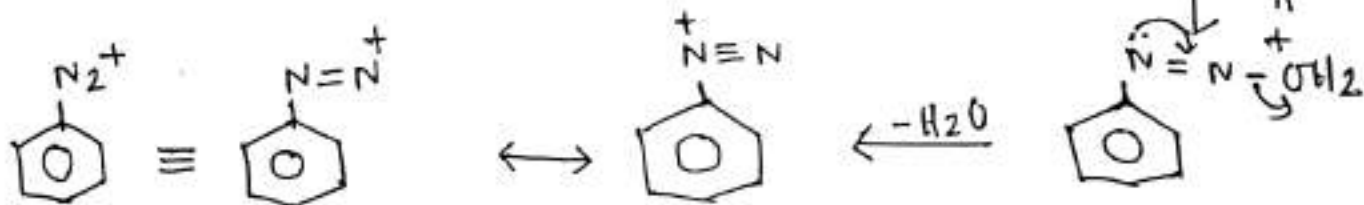
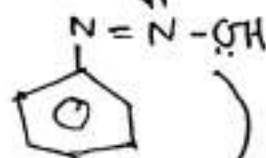
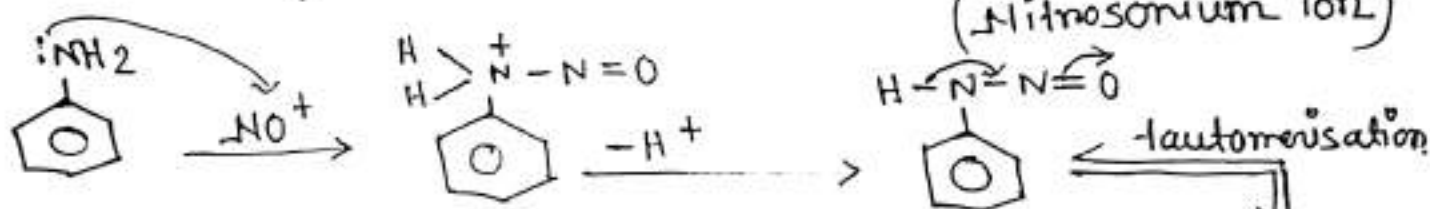
• Trans - 2-amino cyclohexanol $\xrightarrow[\text{dil HCl}]{\text{NaNO}_2}$?



Mechanism of the formation of diazonium ion:-



(Nitrosonium ion)



(Benzene diazonium ion)

Conclusion

Rearrangement reaction is a vast and important topic or chapter in organic chemistry. I

try to speak a little bit of rearrangement reaction which is pinacol - pinacolone rearrangement.

This rearrangement is very helpful for us to synthesise various molecule like spiro compound, carbonyl compound etc.

Thank you

Department of Philosophy

Bidhan Chandra College, Asansol-4

All concern 6th semester Philosophy (Hons .) students are here by informed that the Unit Test 1st assignment, will be held as per following routine in offline mode. The respective students are asked to bring their university Reg. No. for the appearing the test.

Semester	Date	Day	Paper Code	Time	Room no
6 th	19/04/2022	Tuesday	BAHPHIDSE603	12:00-01:00PM	130
6 TH	20/04/2022	Wednesday	BAHPHIC602	12:00-01:00PM	130
6 TH	21/04/2022	Thursday	BAHPHIDSE601	12:00-01:00PM	130
6 th	22/04/2022	Friday	BAHPHIC601	12:00-01:00PM	130

Se. 11.04.22.
Head of the Department
Department of Philosophy
Bidhan Chandra College, Asansol-4

Date: 11/04/2022

Bidhan Chandra College, Asansol.

Unit Test / 6th Sem Hons. - 2022.

Paper: Sadhana (BAHPH/DSE 603)

Dept. of Philosophy.

19.4.22.

5 x 2 = 10

- Q.1. 'अणदि' एर अरुणत क्राइर कर,
- Q.2. 'ज्ञानर शूद्रां अराकृकृतर अनुमोहन' -
दमांर कि कसा(न) शमांर? क्राइर कर.

Department of Commerce

NOTICE

All students of 5th semester of Session 2021-22 Batch are required to submit 'Entrepreneurship Development' Project within 20 December 2021.

Topics :-

- i) Business on Flour Mill
- ii) Tea Blending & Packaging.
- iii) Rural Transformation project.
- iv) Manufacturing Business
- v) A comprehensive group project report on 'Lunch Box delivery service'

Thanking You
Depart. of commerce
29-11-21



Amal

KAZI NAZRUL UNIVERSITY

SESSION: 2021-2022

TOPIC: BUSINESS ON FLOUR MILL

SUBJECT: ENTREPRENEURSHIP DEVELOPMENT

A PROJECT

SUBMITTED TO THE DEPARTMENT OF COMMERCE

BIDHAN CHANDRA COLLEGE

IN PARTIAL FULFILLMENT OF THE REQUIRE FOR THE
BACHELOR OF COMMERCE (PROGRAMME) DEPARTMENT

B.COM (5TH SEMESTER)

SUPERVISED BY

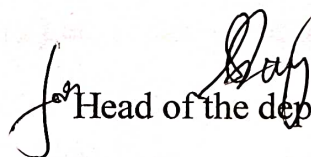
DEEPAK MONDAL

NOTICE

Unit Test of 2nd semester B.Sc. Physics (Honours) will be held as per the following schedule:-

Date & Day	Time	Papers
14/12/2021	11 AM onwards	1. Mathematical methods in Physics 2. Electricity and magnetism

Dated: 07.12.2021



Head of the department

Department of Physics

Bidhan Chandra College

Asansol-4

Head of the Dept. Physics
Bidhan Chandra College
(Govt. Sponsored)
Asansol-713304

Department of Physics
Bidhan Chandra College, Asansol

Second Semester Unit Test Examination

Electricity and Magnetism

Course Code: BSCHPHSC202

F.M. = 20

Time: 1hr.

Answer any four Questions:

5X4 = 20

1. State and explain Thevenin's theorem.
2. Find the potential of a uniformly charged spherical shell of radius R for points (i) inside (ii) outside it using Gauss's Law.
3. Derive expressions for magnetic field, at a point on its axis of a circular coil carrying current.
4. An alternating source of e.m.f. is connected to a circuit having resistance R, capacitance C and inductance L in series. Obtain expression for instantaneous current and impedance. Also obtain the condition of resonance and explain sharpness of resonance.
5. Obtain the generalised expression of Gauss's law in di-electric medium both in integral and differential forms. Discuss the boundary conditions for \vec{D} & \vec{E} across the interface between two media having surface charge density σ

Thevenin's theorem

Any electrical circuit with power sources and resistances can be transformed equivalent theorem's circuit with Thevenin voltage (V_{th}) and ~~emf~~ current (I_{sc}) connected in series with terminals as original circuit.

Steps for calculating V_{th}

1) Make electrical circuit open. Any voltage source ~~change~~ replaced with corresponding internal resistance or in case of ideal voltage sources make ~~emf~~ circuit short. Then take terminal as ground. Now we can calculate thevenin voltage ^{emf} with voltage divider principle. Series, parallel rule. across original circuit terminals.

Steps for calculating I_{sc}

2) Circuit have to open. Any current source can be replaced with corresponding internal resistance ~~emf~~ or in ideal case ~~emf~~ circuit could be made to open. Then state one terminal as ground. Now we can calculate thevenin voltage with voltage divider principle.



By tracing a gaussian surface at distance R to cover full sphere

According to Gauss Law

$$\phi_R \text{ (flux at } R \text{ surface)} = \frac{\text{Charge included}}{\epsilon_0} \quad \leftarrow \text{In free space permittivity}$$

$$\phi_R = \frac{\text{charge over surface } R\text{-shell}}{\epsilon_0}$$

Let charge over shell = Q

$$\therefore \phi_R = \frac{Q}{\epsilon_0} = E \cdot 4\pi R^2$$

$$\therefore E = \frac{Q}{4\pi\epsilon_0 R^2}$$

We know

$$E = -\frac{dV}{dr}$$

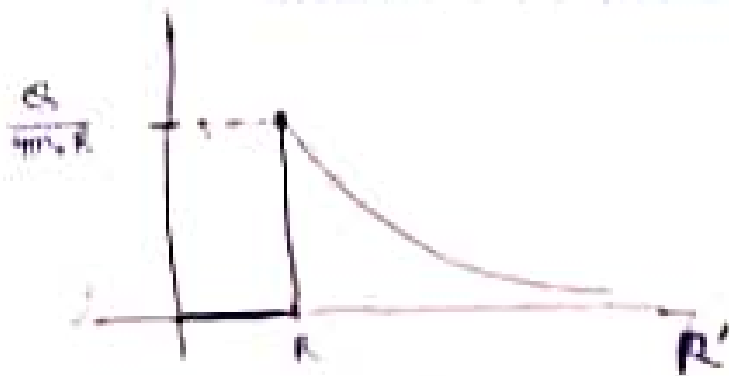
$$\therefore \int_0^V dV = -\int_{\infty}^R E dr$$

$$= \frac{Q}{4\pi\epsilon_0} \int_{\infty}^R \frac{1}{r^2} dr = \frac{Q}{4\pi\epsilon_0 R} = V(R)$$

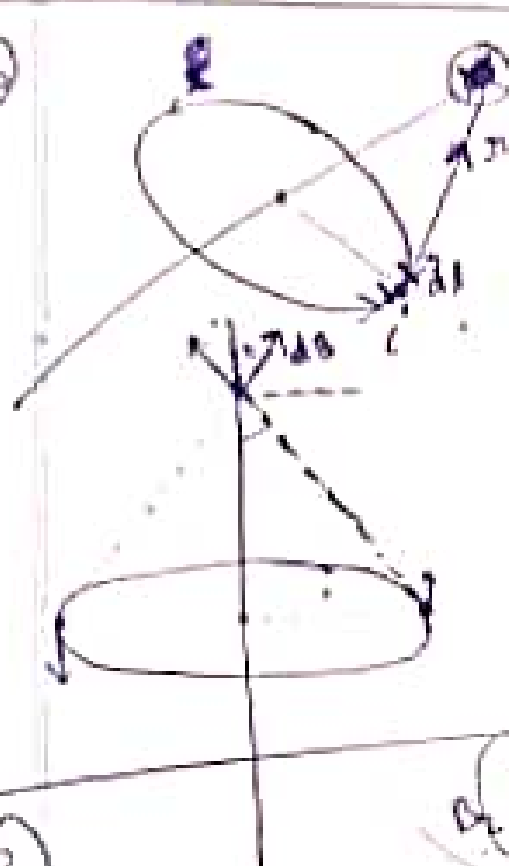
substitution of Electric flux and symmetry problems

at distance r the surface inside have no charge $\therefore \phi_r = 0 \Rightarrow E_r = 0$

$$\therefore V(r) = 0$$



2)

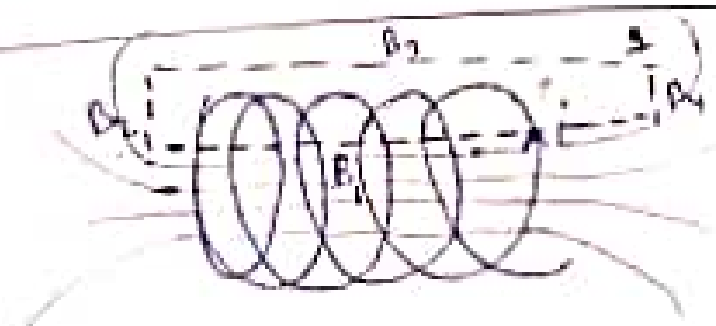


By Biot Savart's Law

$$dB = \frac{\mu_0}{4\pi} \frac{I \, d\vec{l} \times \vec{r}}{r^3}$$

$$dB \cos \theta$$

3)



By Applying Ampere Circuital Law at loop 1

(i) Inside coil magnetic field is approx uniform (B)
 let length of loop is l

$$\therefore (B \cdot l) \text{ is line integral of } (B)$$

(ii) outside coil magnetic field is dispersed, (spread) so, $(B \rightarrow 0)$

$$\therefore \text{the integral of } B \text{ is } (B)$$

(iii) at side way of loop, length is tends to zero \therefore line integral become (0)

$$\therefore B_1 l_1 + B_2 l_2 + B_3 l_3 + B_4 l_4 = \mu_0 i$$

$$B l + B_2 \cdot 0 + 0 \cdot l + B_4 \cdot 0 = \mu_0 i$$

$$B l = \mu_0 i$$

$$\therefore B = \frac{\mu_0 i}{l}$$

For infinite wire, at long distance magnetic field $\therefore B = \frac{\mu_0 i}{2r}$

\Rightarrow Gauss law: Charge enclosed by any arbitrary shape or size of sphere have Electric flux $\phi_E = \frac{Q_{\text{enclosed}}}{\epsilon_0}$

Proof: for Spherical Symmetry



Let's take a imaginary surface a sphere of radius (r) with a charge (q) at its centre.

Electric field at any point of sphere is uniform have same value and always outward Normal to surface

$$\phi_E = \oint \vec{E} \cdot d\vec{S} = \frac{q}{4\pi r^2} \int dS = \frac{q}{4\pi r^2} (4\pi r^2) = \frac{q}{\epsilon_0}$$

$$\therefore \vec{E} \cdot d\vec{S} = E dS$$

$$\therefore \phi_E = \frac{Q}{\epsilon_0} = \oint \vec{E} \cdot d\vec{s} \Rightarrow \oint \vec{D} \cdot d\vec{s} = Q$$

This is true for any other shape or surface, any size and for any position or charge inside surface.

Now $\oint \vec{E} \cdot d\vec{s} = \int \nabla \cdot \vec{E} \, dV$ ← According to vector calculus identity

and let $Q = \int \rho \, dV$

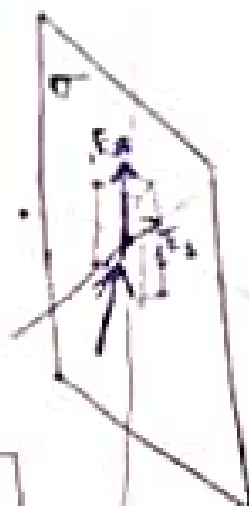
$$\therefore \int \nabla \cdot \vec{E} \, dV = \int \frac{\rho}{\epsilon_0} \, dV$$

$$\therefore \nabla \cdot \vec{E} = \frac{\rho}{\epsilon_0}$$

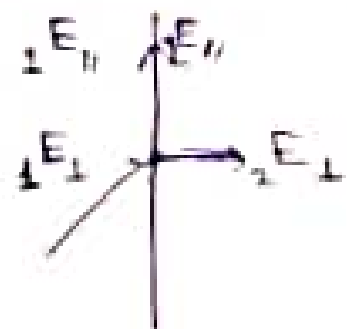
For constant (ϵ) $\nabla \cdot \vec{E} = \frac{\rho}{\epsilon}$

$$\therefore \nabla \cdot \vec{D} = \rho$$

I



II



• Parallel component not involved

$$\therefore (E_{1\perp} - E_{2\perp}) \, dS = \frac{\sigma}{\epsilon} \, dS$$

$$\therefore E_{1\perp} - E_{2\perp} = \frac{\sigma}{\epsilon}$$

$$\vec{D}_{1\perp} - \vec{D}_{2\perp} = \sigma$$